

LESSON PLAN OF 3rd SEMESTER(2023-2024) CHEMICAL ENGINEERING

Discipline :- CHEMICAL	Semester:-3 RD	Name of the Teaching Faculty: Siddhibinayak Pradhan
Subject:- PHYSICAL CHEMISTRY	No of Days/per Week Class Allotted :-04	Semester From: -01 August 2023 To: -30 November 2023
Week	Class Day	Theory/ Practical Topics
1 st	1 st	PHYSICAL PROPERTIES OF LIQUIDS Intermolecular forces in liquid
	2 nd	Vapour pressure and its Effect on Temperature and Boiling point
	3 rd	Surface Tension
	4 th	Viscosity, Measurement of viscosity by Ostwald Method
2 nd	1 st	Refractive Index, specific Refraction
	2 nd	Determination of Refractive index by Refractometer
	3 rd	Optical Activity, measurement of Optical Activity
	4 th	Measurements of Optical Activity
3 rd	1 st	Solved problems based on physical properties of liquids
	2 nd	Chapterwise Test
	3 rd	SOLUTIONS Solution and types of solutions
	4 th	Ways of Expressing concentration
4 th	1 st	Solved numerical related to concentration
	2 nd	Solutions in Gases in Gases
	3 rd	Henry's law and solved problems
	4 th	Solution in liquids in liquids
5 th	1 st	Solubility of partially miscible liquids
	2 nd	Solubility of solid in liquid
	3 rd	Equilibrium concept, solubility curve
	4 th	Raoult's law, ideal solution
6 th	1 st	Explanation of lowering of vapour pressure and its measurements
	2 nd	Concept of elevation of boiling point and depression of freezing point
	3 rd	OSMOSIS AND OSMOTIC PRESSURE Osmosis and Osmotic Pressure with Example
	4 th	Function of semi-permeable Membrane
7 th	1 st	Osmotic pressure and Isotonic pressure
	2 nd	Theories of osmosis
	3 rd	Reverse osmosis
	4 th	The laws of Osmotic Pressure
8 th	1 st	Solved problems on Osmosis

	2nd	Relation between Vapour pressure & Osmotic pressure
	3rd	Relation between Vapour pressure & Osmotic Pressure
	4th	Simple problems
9th	1st	Surprise Test on chapter-1,2,3
	2nd	DISTRIBUTION LAW Introduction
	3rd	Nernst's Distribution Law
	4th	Equilibrium constant from distribution law
10th	1st	Solvent Extraction
	2nd	Multiple Extraction
	3rd	Concept of liquid-liquid Chromatography
	4th	Application of Distribution law
11th	1st	Application of Distribution law
	2nd	Application of Distribution law
	3rd	Numerical problems related to Distribution law
	4th	COLLOIDS Colloids and Types of colloidal system
12th	1st	Characteristics of solutions
	2nd	Applications of colloids
	3rd	Methods of preparation of sols & purifications of sols
	4th	Optical ,kinetic and electrical properties of sols
13th	1st	Emulsion and types of emulsion
	2nd	Roles of Emulsifier
	3rd	Preparation of Emulsions and there properties
	4th	Gel, types of gel,
14th	1st	Properties and Application of gel
	2nd	ADSORPTION Introduction
	3rd	Types of Adsorption
	4th	Physical adsorption and Chemisorption
15th	1st	Application of Adsorption
	2nd	Ion – exchange adsorption
	3rd	Compare absorption and adsorption
	4th	Ion – exchange application.

LESSON PLAN OF 3rd SEMESTER(2024-25) CHEMICAL ENGINEERING

DISCIPLINE: CHEMICAL	Semester:- 3RD	<u>NAME OF THE TEACHING FACULTY</u> Yayati Kishore Mohanta
SUBJECT: FLUID MECHANICS	No of days per Week Allotted : 04	Semester From: -01 August 2023 To: -30 November 2023
Week	Class/ Day	Theory/ Practical Topics
1st	1st	Units And Dimensions; Fluid and its classification
	2nd	Properties of fluid and its units
	3rd	Newton's law of viscosity
	4th	Newtonian & Non-Newtonian fluid
2nd	1st	Hydrostatic equilibrium and pressure head
	2nd	Fluid pressure measuring devices
	3rd	Different types of manometers and its applications
	4th	Derivation of manometric equation
3rd	1st	Problems on Manometric Equation
	2nd	Equation of continuity
	3rd	Problems on Continuity Equation
	4th	Types of fluid flow
4th	1st	Laminar and turbulent flow
	2nd	Reynolds's number, critical velocity
	3rd	Mechanism of fluid flow in pipes
	4th	Reynolds' experiment
5th	1st	Bernoulli's theorem, pump work (solve simple problems)
	2nd	Bernoulli's theorem, pump work (solve simple problems)
	3rd	Bernoulli's theorem, pump work (solve simple problems)
	4th	Flow of incompressible fluids in pipe

6th	1st	Flow of incompressible fluids in pipe
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	2nd	Flow of incompressible fluids in pipe
	3rd	Friction factor, roughness
	4th	Estimate friction loss in pipes & coils, equivalent length
7th	1st	Fanning's equation (Solve simple problems)
	2nd	Fanning's equation (Solve simple problems)
	3rd	Friction losses through sudden enlargement in pipes
	4th	Friction losses through sudden contraction in pipes
8th	1st	Problems on friction losses through sudden enlargement in pipes
	2nd	Problems on friction losses through sudden contraction in pipes
	3rd	Flow of fluids in non-circular conduits. Water hammer
	4th	Working of flow measuring devices, advantages & disadvantages
9th	1st	Expression for flow measurement through orifice meter
	2nd	Expression for flow measurement through venturi meter
	3rd	Expression for flow measurement through pitot tube
	4th	Working of Rota meter and its calibration
10th	1st	Simple problems on flow measurement
	2nd	Simple problems on flow measurement
	3rd	Simple problems on flow measurement
	4th	Simple problems on flow measurement
11th	1st	Concept of transportation of fluid by pipes and tubes
	2nd	Different pipe fittings and its application
	3rd	Different types of valves and their applications
	4th	Classification of pumps
12th	1st	Construction and working of centrifugal pump
	2nd	Performance characteristics of centrifugal pumps
	3rd	Cavitation, Net positive suction head, Air binding & priming of pump

	4th	Centrifugal pump troubles and remedies
13th	1st	Construction and working of centrifugal pump
	2nd	Performance characteristics of centrifugal pumps

	3rd	Working of Piston pump, plunger pump, gear pump, diaphragm pump
	4th	Pumping device for gas: blower, compressor and vacuum devices
14th	1st	Pressure drop in porous medium
	2nd	Concept of fluidization
	3rd	Types of fluidization
	4th	Minimum fluidization velocity
15th	1st	Fluidized bed pressure drop
	2nd	Principle of pneumatic conveyance
	3rd	Flow through packed bed; Problems on fluidisation
	4th	Previous Year Questions Practice

LESSON PLAN OF 3rd SEMESTER (2023-24) CHEMICAL ENGINEERING DEPARTMENT		
Discipline: Chemical	Semester: 3rd	Name of The Teaching Faculty: Siddhibinayak Pradhan
Subject: Theory-3 Mechanical Operation	No of Days per week class allotted:4	Semester From: -01 August 2023 To: -30 November 2023 No of Weeks: 15
Week	Class days	Theory/Practical Topic
1 st	1 st	Introduction to mechanical operation
	2 nd	Objectives of size reduction
	3 rd	State laws of crushing like Bonds law, Rittinger's law, Kick's law
	4 th	Practice numerical related to different laws
2 nd	1 st	Concept of Crushing efficiency, Work index
	2 nd	Classification of size reduction equipment and their construction and operation
	3 rd	Construction and working of Jaw crusher, Gyratory crusher
	4 th	Construction and working of Smooth roll crusher, Hammer Mill, Ball Mill
3 rd	1 st	Closed and open circuit grinding, dry grinding
	2 nd	Define wet grinding, free and choke grinding
	3 rd	Practice of the chapter and solve simple problems
	4 th	Practice previous years questions
4 th	1 st	Objectives of size separation
	2 nd	Shape and size of irregular particle
	3 rd	Different types of screen analysis
	4 th	Define ideal screen & actual screen
5 th	1 st	Material balance over the screen
	2 nd	Construction and operation of different types of industrial screens and their effectiveness
	3 rd	Construction and operation of different types of industrial screens and their effectiveness
	4 th	Construction and operation of air filters, air separator
6 th	1 st	Construction and working of cyclone separator, magnetic and Electromagnetic separation
	2 nd	Theory of settling like free

7th	3rd	Hindered settling
	4th	State Stoke's law, Classification
	1st	Solve simple numerical based on the chapter
	2nd	Theory on Sedimentation, thickeners, clarifiers

	3 rd	Theory on hydraulic classifiers, jigs, classifier riffled table and their use
	4 th	Principle & operation of froth floatation and its use
8 th	1 st	Revision of the chapter and practice previous year question
	2 nd	Types of filtrations, Theory of filtration,
	3 rd	Types of cakes, cake resistance, pressure drop, filter medium
	4 th	Filter Aids and related derivation
9 th	1 st	Classification, constructions and working principles of filtration equipments, Thickeners
	2 nd	Classification, constructions and working principles of filtration equipments, Thickeners
	3 rd	Batch and continuous centrifuges with their construction, operation
	4 th	Uses of batch and continuous centrifuges
10 th	1 st	Flocculation, coagulants and role of coagulant in filtration
	2 nd	Practice questions based on the chapter
	3 rd	Doubt clearing class
	4 th	Theory on mixing
11 th	1 st	Various mixing operations like Mixing of liquid with liquid
	2 nd	Mixing of liquid with solid
	3 rd	Mixing of viscous materials
	4 th	Mixing of Solid with solid
12 th	1 st	Mixing of gases with liquids
	2 nd	The flow pattern in agitated vessel
	3 rd	Methods of prevention of swirling and vortex formation, baffling
	4 th	Different impellers used in mixing operation
13 th	1 st	Different propellers, paddles used in mixing operation
	2 nd	Revision of the chapter and practice different questions
	3 rd	Introduction to transportation and storage
	4 th	Objectives of transportation and storage
14 th	1 st	Transportation of solid by belt conveyor
	2 nd	apron conveyor, screw Conveyor
	3 rd	bucket elevators, scrapers and pneumatic conveyers
	4 th	Storage and handling of solids
15 th	1 st	construction and uses of silos and bins
	2 nd	Revision of the chapters
	3 rd	Doubt clearing class

	4th	Practice question answer
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**LESSON PLAN OF CHEMICAL ENGINEERING DEPARTMENT WINTER
2024**

Discipline: Chemical	Semester: 3rd	Name of Faculty: Sibasish Mahapatra
Subject: Industrial Stoichiometry (TH-4)	No of Days per week class allotted:4	Semester From: -01 August 2023 To: -30 November 2023
Week	Class Day	Theory Topics
1 st	1 st	CHAPTER-1: UNITS AND DIMENSIONS Introduction
	2 nd	Basic and derived units used in process industry.
	3 rd	Units of physical and chemical properties
	4 th	Relation between units and dimension
2 nd	1 st	Unit conversion and solve numerical
	2 nd	Concepts of unit operation and unit process
	3 rd	Application of various graphs in process calculation
	4 th	Solve numerical
3 rd	1 st	CHAPTER-2: MOLE CONCEPT Atomic number, atomic weight of elements
	2 nd	Mol. Wt., mole unit, mole fraction (or percent) and mass fraction (or percent),
	3 rd	Relation between mole and mass fraction
	4 th	Mole concept with respect to chemical equation.
4 th	1 st	Principle of atom conservation.
	2 nd	Mole calculation from reaction
	3 rd	Methods of expressing composition of mixtures and solutions
	4 th	Solve related numericals
5 th	1 st	CHAPTER-3: STOICHIOMETRY Introduction
	2 nd	Concept of limiting reactant, Atomic weight,
	3 rd	Concept of Molecular weight and empirical formula
	4 th	Solved numerical based on limiting reactant, mass-mass and mass volumebasis
6 th	1 st	Concepts of Eq. weight, valence of molecule
	2 nd	Solve related numerical
	3 rd	Concepts of preparation of solution
	4 th	Weight and volume percent of solutions
7 th	1 st	Basics of Normality, molarity and molality
	2 nd	Numerical on solution preparation
	3 rd	Solve related numerical
	4 th	CHAPTER-4: GASES AND GASEOUS MIXTURES

		Introduction
8th	1st	Define gases, different gaseous mixture
	2nd	Derivation of Ideal gas equation
	3rd	Derive average molecular weight and Values of R
	4th	Derivation of density of gas mixture
9th	1st	Solve related numerical
	2nd	Composition by vol% and by weight % related to average molecularweight of gas mixture
	3rd	Solve the examples and exercises related to Avg. mol wt. and Ideal gasequation.
	4th	Concepts of Pressure, partial pressure and various laws related toPVT behavior.
10th	1st	Concepts of State Raoult's law and Henry's law
	2nd	CHAPTER-5: MATERIAL BALANCE WITHOUT CHEMICAL REACTION Introduction
	3rd	Basics of chemical equation and stoichiometry
	4th	Concepts of law of conservation of mass and material balance over thereaction.
11th	1st	Material balance problems without chemical reactions of unit operations
	2nd	Material balance of Evaporation and solve numerical
	3rd	Material balance of mixing and solve numerical
	4th	Material balance of crystalization
12th	1st	Material balance over distillation and solve numerical
	2nd	Material balance over drying and solve related numerical
	3rd	Material balance humidification and solve related numerical
	4th	Material balance over filtration
13th	1st	Material balance over absorption, extraction
	2nd	Solve numerical
	3rd	CHAPTER-6: MATERIAL BALANCE WITH CHEMICALREACTION Introduction
	4th	Concepts of Limiting reactant, Excess reactant
14th	1st	Concepts of Conversion, Selectivity, Yield.
	2nd	Basic concepts involved in material balance calculations.
	3rd	Material balance over combustion
	4th	Material balance over chemical reaction calculation
15th	1st	Concepts of heat of combustion and heat of formation.
	2nd	Concept of recycle and by pass, purge
	3rd	Excess air and theoretical air
	4th	Numerical based on combustion, Excess air and theoretical air

**LESSON PLAN OF CHEMICAL ENGINEERING DEPARTMENT WINTER
2024**

Discipline: Chemical		Semester: 3rd	Name of Faculty: Satya Sankar Raj	
Subject: Theory-5 Environmental Studies		No of Days per week class allotted	Semester From: -01 August 2023 To: -30 November 2023	
Week	Class No	Class days	Chapter	Theory Topic
			Chapter -1 The Multidisciplinary nature of environmental studies	
1 st	1	1 st	1.1	Definition & important issues with environmental science
	2	2 nd	1.2	Scope and importance of environmental studies
	3	3 rd	1.3	Components of Environment and its importance
	4	4 th	1.4	Need for public awareness, Institution related to environmental studies
			Chapter -2 Natural Resources, Renewable and nonrenewable resource	
2 nd	5	1 st	2.1.1	Forest resources: Use and over-exploitation, deforestation, case studies,
	6	2 nd	2.1.1	Timber extraction, mining, dams and their effects on forests and tribal people
	7	3 rd	2.1.2	Water resources: Use and over-utilization of surface and ground water, floods,
	8	4 th	2.1.2	Drought, conflicts over water, dam's benefits, and problems
3 rd	9	1 st	2.1.3	Mineral Resources: Use and exploitation, environmental effects of extracting
	10	2 nd	2.1.4	Food Resources: World food problems, changes caused by agriculture
	11	3 rd	2.1.4	Effects of modern agriculture, fertilizers- pesticides problems, water logging
	12	4 th	2.1.5	Energy Resources: Growing energy need, renewable and non-renewable
4 th	13	1 st	2.1.6	Land Resources: Land as a resource, soil erosion, and desertification
	14	2 nd	2.2-2.3	Role of individual in conservation of natural resources, sustainable lifestyles
			Chapter -3 Systems	
	15	3 rd	3.1	Concept of an eco-system, understanding Eco system, Resource Utilization
	16	4 th	3.2	Structure and function of an eco-system- Structural & functional aspects
5 th	17	1 st	3.3	Producers, consumers, decomposers- Examples in eco system
	18	2 nd	3.4	Energy flow in the eco system- examples of different cycles
	19	3 rd	3.5	Ecological succession-examples in eco system
	20	4 th	3.6	Food chains, food webs and ecological pyramids □
6 th	21	1 st	3.7	Introduction characteristic function of eco system: Forest ecosystem
	22	2 nd	3.8	Aquatic eco systems (ponds, streams, lakes, rivers, oceans, estuaries)
			Chapter -4 Biodiversity and it's Conservation	
	23	3 rd	4.1	Introduction-Definition: genetics, species, and ecosystem diversity
	24	4 th	4.2	Biogeographically classification of India

Week	Class No	Class days	Chapter	Theory Topic
7 th	25	1 st	4.3	Value of biodiversity: consumptive use, productive use
	26	2 nd	4.3	Social ethical, aesthetic and Option values
	27	3 rd	4.4	Biodiversity at global, national, and local level
	28	4 th	4.5	Threats to biodiversity: Habitats loss, Hot spot of biodiversity
8 th	29	1 st	4.5	poaching of wildlife, man wildlife conflicts, India as megadiversity nation
	30	2 nd	4.5	Conservation of biodiversity-In situ Conservation, Ex situ Conservation
			Chapter -5 Environmental Pollution	
	31	3 rd	5.1.1	Definition Causes, effects, and control measures of air pollution
	32	4 th	5.1.2	Definition Causes, effects, and control measures of water pollution
9 th	33	1 st	5.1.3	Definition Causes, effects, and control measures of soil pollution
	34	2 nd	5.1.4	Definition Causes, effects, and control measures of marine pollution
	35	3 rd	5.1.5	Definition Causes, effects and control measures of noise pollution
	36	4 th	5.1.5	Definition Causes, effects and control measures of thermal pollution
10 th	37	1 st	5.1.6	Definition Causes, effects and control measures of nuclear hazards
	38	2 nd	5.2	Solid waste Management: Causes, effects, and management
	39	3 rd	5.2	Control measures of urban and industrial wastes
	40	4 th	5.3	Role of an individual in prevention of pollution
11 th	41	1 st	5.4	Disaster management: Floods, earthquake and its mitigation measure
	42	2 nd	5.4	cyclone and landslides its mitigation measure
			Chapter -6 Social Issues and the Environment	
	43	3 rd	6.1	Form unsustainable to sustainable development
	44	4 th	6.2	Urban problems related to energy
12 th	45	1 st	6.3	Water conservation, rainwater harvesting, water shed management
	46	2 nd	6.4	Resettlement and rehabilitation of people; its problems and its concern
	47	3 rd	6.5	Environmental ethics: issue and possible solutions
	48	4 th	6.6	Climate change, global warming, acid rain, ozone layer depletion,
13 th	49	1 st	6.6	Nuclear accidents and holocaust, case studies
	50	2 nd	6.7	Air (prevention and control of pollution) Act
	51	3 rd	6.8	Water (prevention and control of pollution) Act
	52	4 th	6.9	Public awareness, Issues related to Environment legislation
			Chapter -7 Human population and the environment	
14 th	53	1 st		Population growth and variation among nations
	54	2 nd		Population explosion- family welfare program
	55	3 rd		Environment and human health, Environmental Health, Climate health

Week	Class No	Class days	Chapter	Theory Topic
	56	4 th		Human rights, issues connected with environment and human rights
15 th	57	1 st		Value education, Environmental value, valuing nature, valuing culture
	58	2 nd		Social Justice, Human heritage, Equitable use of resources
	59	3 rd		Common Property resources, Equitable use of resources
	60	4 th		Role of information technology in environment and human health

LESSON PLAN OF 3 rd SEMESTER (2023-24) CHEMICAL ENGINEERING DEPARTMENT		
Discipline: Chemical	Semester: 3rd	Name of The Teaching Faculty: Yayati Kishore Mohanta
Subject: Practical 5 Chemical Engg. Drawing	No of Days per week class allotted:4	Semester From: -01 August 2023 To: -30 November 2023
Week	Practical days	Practical Topic
1 st	1 st	1.1 Draw symbols of equipment used in chemical industries
	2 nd	Draw symbols of equipment used in chemical industries
	3 rd	1.2 Draw symbol of pipe line
	4 th	Valves
2 nd	1 st	Pumps
	2 nd	Compressor
	3 rd	heating cooling arrangements
	4 th	Furnaces
3 rd	1 st	Boilers
	2 nd	Practice symbols
	3 rd	process vessels
	4 th	Storage vessels
4 th	1 st	Driers
	2 nd	Separators
	3 rd	Filters
	4 th	Centrifuge
5 th	1 st	Stirrer
	2 nd	Feeder
	3 rd	Conveyor
	4 th	Practice all the symbols
6 th	1 st	2. Draw sketch of Chemical Engineering Equipment like Heat exchanger (double pipe and shell)
	2 nd	Heat exchanger (tube type)
	3 rd	Distillation column

	4 th	Dryer
7 th	1 st	Evaporator
	2 nd	Ball mill
	3 rd	Practice the chemical engineering equipments
	4 th	Cyclone Separator
8 th	1 st	Crystalliser
	2 nd	Absorber
	3 rd	Extractor
	4 th	3.1 PROCESS INSTRUMENTATION DIAGRAM Draw symbols of flow rate indicator
9 th	1 st	flow recorder
	2 nd	Draw symbols of level indicator
	3 rd	pH recorder
	4 th	level controller
10 th	1 st	Class test related to symbols
	2 nd	3.2 Draw P.I. diagrams of Cooler temperature control
	3 rd	Reactor temperature control
	4 th	Heater temperature control
11 th	1 st	Hot fluid temperature control
	2 nd	Practice different control system
	3 rd	Evaporator circulation control
	4 th	Evaporator circulation control
12 th	1 st	Tray dryer control
	2 nd	Class test
	3 rd	Top temperature control of distillation column
	4 th	Top temperature control of distillation column
13 th	1 st	Control of level
	2 nd	reflux condenser
	3 rd	Steam flow rate
	4 th	level control of reboiler
14 th	1 st	Practice previous diagram
	2 nd	Class test

15 th	3 rd	3.3 Utility Line diagram Service fluid code for piping
	4 th	Utility block diagram for steam
	1 st	Utility block diagram for chilled water
	2 nd	Practice the diagram
	3 rd	Practice the diagram
	4 th	Practice the diagram

Discipline: Chemical		Semester: 3 rd	Name of Faculty: All Faculty	
Subject: Student Centric Activity		No of Days per week class allotted-03	Semester From: -01 August 2023 To: -30 November 2023	
Week	Class No	Class days		
1 st	1	1 st	September 4 th Week	Orientation Program and Mentor Mentee Meet
	2	2 nd		
	3	3 rd		
2 nd	4	1 st	October 2 nd week	Poster Making on Emerging trends in different Chemical Industry/Energy Conservation/
	5	2 nd		
	6	3 rd		
3 rd	7	1 st	October 3 rd week	Seminar by Industry Expert- latest trend in Plastic Processing- Proprietor Sree Plast Limited or Functioning of State Pollution Control Board- RO Regional Office OSPCB
	8	2 nd		
	9	3 rd		
4 th	10	1 st	October 4 th week	Laboratory Maintenance- 1.Cleaning of equipment,2. Lubrication 3. Running of equipment 4.Removal of residue material 5.Painting of parts, 6.Arranging glass ware, Chemicals 7. Minor maintenance of equipment
	11	2 nd		
	12	3 rd		
5 th	13	1 st	November 1 st week	Creativity & Idea Presentation-
	14	2 nd		
	15	3 rd		
6 th	16	1 st	November 2 nd week	Seminar by Industry Expert- Pharmaceutical Intermediate Processing-Dept of Pharmacy BU/ Roland Institute of pharmacy
	17	2 nd		
	18	3 rd		
7 th	19	1 st	November 4 th week	Field Visit or Industry visit- JK paper/ Waste Treatment plant Mahuda/SreePlast limited
	20	2 nd		
	21	3 rd		
8 th	22	1 st	December 1 st week	CV/ Interview preparation/Career Counseling Program
	23	2 nd		
	24	3 rd		
9 th	25	1 st	December 2 nd week	Laboratory Maintenance-1.Cleaning of equipment,2. Lubrication 3. Running of equipment 4.Removal of residue material 5.Painting of parts, 6.Arranging glass ware, Chemicals 7. Minor maintenance of equipment
	26	2 nd		
	27	3 rd		
10 th	28-30	1 st - 3 rd	December 3 rd week	Seminar by Industry Expert- From IISER/ CoE BU in the latest area of research

LESSON PLAN OF 3 rd SEMESTER (2023-24) CHEMICAL ENGINEERING DEPARTMENT		
Discipline: Chemical	Semester: 3rd	Name of The Teaching Faculty: Sibasish Mahapatra
Subject: Practical-3 Mechanical Operation	No of Days per week class allotted:3	Semester From: -01 August 2023 To: -30 November 2023
Week	Practical days	Practical Topic
1 st	1 st	Demonstrate operation of a Blake type jaw crusher and Verify Rittinger's Law and the capacity of jaw crusher
	2 nd	
	3 rd	
2 nd	1 st	a. Demonstrate operation of a Ball mill b. Find-out the critical speed of a ball mill and compare with the actual speed
	2 nd	
	3 rd	
3 rd	1 st	Determine the effect the number of balls and time of grinding and plot a graph between the no. of balls Vs. Time
	2 nd	
	3 rd	
4 th	1 st	Demonstrate operation of sieve shaker
	2 nd	
	3 rd	
5 th	1 st	determine the average size of the product after performing separation size separation by screen analysis
	2 nd	
	3 rd	
6 th	1 st	Demonstrate operation of vibrating screen & find-out its screen efficiency
	2 nd	
	3 rd	
7 th	1 st	Demonstrate operation of froth flotation cell and Concentrate the given coal sample and find out the ash present after and before concentration
	2 nd	
	3 rd	
8 th	1 st	Perform the batch sedimentation test and plot a graph between height of the dead zone and time
	2 nd	
	3 rd	

9 th	1 st	Demonstrate operation of a magnetic separator
	2 nd	
	3 rd	
10 th	1 st	Demonstrate operation of cyclone separator and estimate its efficiency
	2 nd	
	3 rd	
11 th	1 st	Demonstrate operation of Wilflay table
	2 nd	
	3 rd	
12 th	1 st	Demonstrate operation of a centrifuge
	2 nd	
	3 rd	
13 th	1 st	Demonstrate operation of a classifier
	2 nd	
	3 rd	
14 th	1 st	Determine operation of a paddle mixer
	2 nd	
	3 rd	
15 th	1 st	Demonstrate operation of filter press
	2 nd	
	3 rd	

Discipline: Chemical		Semester: 3rd	Name of Faculty: Satya Sankar Raj	
Subject: Practical-1 Physical Chemistry Laboratory		No of Days per week class allotted-04	Semester From: -01 August 2023 To: -30 November 2023	
Week	Class No	Class days	Chapter	Practical Topic
			Experiment no-1 Preparation of standard solution of an acid and alkali	
1 st	1	1 st	1.1	To acquaint with safety and SOP of Chemistry laboratory
	2	2 nd	1.2	To understand the concept of Normality, Molarity and standard solution
	3	3 rd	1.3	Demonstration of experiment
	4	4 th	1.4	Practice on preparation of standard solution
2 nd	5	1 st	1.4	Practice on preparation of standard solution
	6	2 nd	1.4	Determine the strength of standard solution
			Experiment no-2 Determine the viscosity of a liquid by Red wood viscometer at different temperatures and plotting graph between viscosity and temperature	
	7	3 rd	2.1	Concept of viscosity and theory of viscosity measurement
	8	4 th	2.2	Description of Viscometer and demonstration of working of viscometer
3 rd	9	1 st	2.3	Demonstration of experiment
	10	2 nd	2.4	Practice on Determination of viscosity of standard oil
	11	3 rd	2.4	Practice on preparation of viscosity of standard oil at different temperature
	12	4 th	2.5	Calculation of viscosity and plotting of graph
			Experiment no-3 To determine the partition coefficient of iodine between water and carbon tetrachloride at room temperature	
4 th	13	1 st	3.1	Study of solubility of solvent in different solvent and its behavior
	14	2 nd	3.2	Nernst's distribution law- statement and explanation, limitation
	15	3 rd	3.3	Demonstration of solvent extraction
	16	4 th	3.4	Determine the partition coefficient of iodine between water and CCl ₄
5 th	17	1 st	3.4	Practice on experimental technique of solvent extraction
	18	2 nd	3.5	Calculation, Result, and discussion on partition coefficient of iodine
			Experiment no-4 To determine the partition coefficient of benzoic acid between water and benzene at room temperature and molecular state of Benzoic acid in benzene as compared to its solution in water.	
	19	3 rd	4.1	Study of behavior of benzoic acid between water and benzene
	20	4 th	4.2	Demonstration of Experiment on solvent extraction
Week	Class No	Class days	Chapter	Practical Topic
6 th	21	1 st	4.3	Determine partition coefficient of Benzoic acid between water and Benzene

	22	2 nd	4.4	Practice on experimental technique of solvent extraction
	23	3 rd	4.4	Practice on experimental technique of solvent extraction
	24	4 th	4.5	Calculation, Result, and discussion on partition coefficient of benzoic acid
			Experiment no-5 To prepare colloidal solution of starch.	
7 th	25	1 st	5.1	Study on Colloidal state, colloid and types of colloidal system
	26	2 nd	5.2	Study on classification of colloids
	27	3 rd	5.3	Demonstration of Preparation of colloidal solution of starch
	28	4 th	5.4	Preparation of colloidal solution of starch
8 th	29	1 st	5.4	Practice on Preparation of colloidal solution of starch
	30	2 nd	5.5	Result, and discussion, application on colloidal system
			Experiment no-6 To prepare colloidal solution of egg albumin	
	31	3 rd	6.1	Study of characteristic of lyophilic sol and lyophobic sol
	32	4 th	6.2	Discussion on different methods of preparation of sols, purification of sols
9 th	33	1 st	6.3	Demonstration of Preparation of colloidal solution of starch
	34	2 nd	6.4	Preparation of colloidal solution of starch
	35	3 rd	6.4	Practice on Preparation of colloidal solution of starch
	36	4 th	6.5	Result, and discussion, application on colloidal system
			Experiment no-7 Determine the solubility of a given salt at room temperature and draw its solubility curve.	
10 th	37	1 st	7.1	Study of solution, solubility, Types, Solution of solids in liquids
	38	2 nd	7.2	Discussion on determination of solubility and solubility curve
	39	3 rd	7.3	Demonstration of the experiment on solubility of a given salt
	40	4 th	7.4	Practice on determine solubility at different temperature
11 th	41	1 st	7.4	Practice on determine solubility at different temperature
	42	2 nd	7.5	Result, and discussion, application of solubility curve
			Experiment no-8 To determine the adsorption isotherm of acetic acid by activated charcoal.	
	43	3 rd	8.1	Study of Principle of adsorption, type, Comparison
	44	4 th	8.2	Study of adsorption isotherm, plotting, limitations, and assumption
12 th	45	1 st	8.3	Demonstration of the experiment on adsorption of acetic acid by charcoal.
	46	2 nd	8.4	Practice on adsorption of acetic acid by charcoal
	47	3 rd	8.4	Practice on adsorption of acetic acid by charcoal
	48	4 th	8.5	Result, and discussion, application of adsorption isotherm of acetic acid
Week	Class No	Class days	Chapter	Practical Topic

			Experiment no-9 To investigate the adsorption of oxalic acid from aqueous solution of activated charcoal and examines the validity of Freundlich and Langmuir's adsorption isotherm	
13 th	49	1 st	9.1	Study of characteristic of Freundlich and Langmuir's adsorption isotherm
	50	2 nd	9.2	Study of plotting of Freundlich and Langmuir's adsorption isotherm
	51	3 rd	9.3	Demonstration of the experiment on adsorption of oxalic acid by charcoal.
	52	4 th	9.4	Practice on adsorption of oxalic acid by charcoal
14 th	53	1 st	9.4	Practice on adsorption of oxalic acid by charcoal
	54	2 nd	9.5	Result, and discussion, application of adsorption isotherms of oxalic acid
			Experiment no-10 To determine the rate constant for hydrolysis of ethyl acetate catalyzed by hydrochloric acid	
	55	3 rd	10.1	
	56	4 th	10.2	
15 th	57	1 st	10.3	Demonstration of the experiment on hydrolysis of ethyl acetate.
	58	2 nd	10.4	Practice on hydrolysis of ethyl acetate
	59	3 rd	10.4	Practice on hydrolysis of ethyl acetate at different time interval
	60	4 th	10.5	Result, and discussion, application of adsorption isotherms of oxalic acid

**LESSON PLAN OF CHEMICAL ENGINEERING DEPARTMENT WINTER
2024**

Discipline: Chemical	Semester: 5th	Name of Faculty: Yayati Kishore Mohanta/Siddhibinayak Pradhan
Subject: Practical-4 Environmental Engineering Laboratory	No of periods per week allotted:3	Semester From: -01 August 2023 To: -30 November 2023
Week	Experiment	Experiment Topic
1 st	1	Collection of sample of waste water
2 nd	2	Analyze a given sample of waste water for estimation of dissolved chloride
3 rd	2	Analyze a given sample of waste water for estimation of dissolved chloride
4 th	3	Determine the dissolved oxygen content of water by Winkler's method
5 th	3	Determine the dissolved oxygen content of water by Winkler's method
6 th	4	Determine the chemical oxygen demand (BOD) exerted by a given sample of waste water
7 th	4	Determine the chemical oxygen demand (BOD) exerted by a given sample of waste water
8 th	5	Determine the chemical oxygen demand (COD) of a given sample of waste water
9 th	5	Determine the chemical oxygen demand (COD) of a given sample of waste water
10 th	6	Determine the turbidity of a given sample of waste water
11 th	7	Determine the total dissolved solid in a given sample of waste water
12 th	7	Determine the total dissolved solid in a given sample of waste water
13 th	8	Determine the optimum amount of Coagulant required to treat to turbid water
14 th	8	Determine the optimum amount of Coagulant required to treat to turbid water
15 th	9	Determine the amount of sulphate in a given sample of water.

Discipline:Chemical	Semester:3rd	Name of Faculty: Yayati Kishore Mohanta	
Subject: Practical-2 Fluid mechanics Engg. Laboratory	No of periods per week allotted:3	Semester From: -01 August 2023 To: -30 November 2023	
Week	Practical days	Experiment	Experiment Topic
1 st	1 st	1	Demonstrate operation of different types of manometers
2 nd	1 st	2	Demonstrate operation of Reynolds's apparatus and find out critical velocity
3 rd	1 st	2	Demonstrate operation of Reynolds's apparatus and find out critical velocity
4 th	1 st	3	Verify Bernoulli's equation
5 th	1 st	3	Verify Bernoulli's equation
6 th	1 st	4	Demonstrate operation of venturimeter and determine the venturi co-efficient
7 th	1 st	5	Demonstrate operation of orificemeter and determine the orifice co-efficient
8 th	1 st	6	Determine co-efficient of friction and friction losses in pipes
9 th	1 st	7	Demonstrate operation of a Rota meter and determine rate of flow through rota meter
10 th	1 st	8	Demonstrate the flow over a weir or 'V' notch
11 th	1 st	9	Demonstrate operation of a fluidized bed column
12 th	1 st	10	Demonstrate operation of a centrifugal pump and valves
13 th	1 st	11	Demonstrate operation of Helical Coil and determine pressure drop through the coil.
14 th	1 st	12	Basic plumbing practice
15 th	1 st	12	Basic plumbing practice