

Question Bank for Mass Transfer - II

Chapter 1: Humidification and Dehumidification	
1.	Define wet bulb and dry bulb temperatures and explain their significance in humidification processes. (L1 – Remember, CO1)
2.	Explain the principle of wet bulb temperature theory with examples. (L2 – Understand, CO1)
3	Differentiate between absolute humidity and relative humidity with calculations. (L3 – Apply, CO1)
4	Describe the construction and working of natural and mechanical draft cooling towers with diagrams. (L2 – Understand, CO2)
5	Analyze the performance of a cooling tower based on given data. (L4 – Analyze, CO4)
6.	Solve a numerical problem to determine the water loss due to evaporation in a cooling tower. (L4 – Analyze, CO4)
6	Using a psychrometric chart, calculate the specific humidity and enthalpy of air at a given condition. (L3 – Apply, CO5)
7	Explain the methods of humidification and dehumidification and their industrial applications. (L2 – Understand, CO2)
8	Derive the equation for heat and mass transfer in a cooling tower. (L3 – Apply, CO4)
9	Define wet bulb and dry bulb temperatures and explain their significance in humidification processes. (L1 – Remember, CO1)
Chapter 2: Drying	
1.	Define drying and classify the types of moisture present in solids. (L1 – Remember, CO1)
2.	Explain the constant rate and falling rate periods of drying with a drying rate curve. (L2 – Understand, CO2)
3.	Differentiate between free moisture and bound moisture with examples. (L3 – Apply, CO2)
4.	Describe the construction and working principle of a rotary dryer and fluidized bed dryer. (L2 – Understand, CO3)
5.	Explain the mechanism of drying in a spray dryer and its application in food industries. (L2 – Understand, CO2)
6.	Solve a problem to calculate the drying time for a batch process given the initial and final moisture content. (L4 – Analyze, CO4)
7.	Perform an energy balance calculation for a tray dryer to determine the heat requirement. (L4 – Analyze, CO4)
8	Compare the working principles of tunnel dryer and flash dryer with examples. (L3 – Apply, CO2)
9	Explain the choice of dryer for heat-sensitive materials with industrial examples. (L2 – Understand, CO3)
Chapter 3: Extraction	

1.	Define liquid-liquid extraction and explain its significance in chemical engineering. (<i>L1 – Remember, CO1</i>)
2.	Explain the principle of solid-liquid extraction with examples. (<i>L2 – Understand, CO1</i>)
3.	Differentiate between batch and continuous extraction methods with industrial applications. (<i>L3 – Apply, CO2</i>)
4.	Describe the construction and working of a mixer-settler used in liquid-liquid extraction. (<i>L2 – Understand, CO3</i>)
5.	Solve a numerical problem to determine the solvent-to-feed ratio in a liquid-liquid extraction process. (<i>L4 – Analyze, CO4</i>)
6	Explain the parameters affecting the choice of solvent in liquid-liquid extraction. (<i>L2 – Understand, CO2</i>)
7	Derive the material balance equation for a counter-current solid-liquid extraction system. (<i>L3 – Apply, CO4</i>)
8	Perform a calculation to determine the extraction efficiency for a given system using equilibrium data. (<i>L4 – Analyze, CO5</i>)
9	Describe the design and working of a continuous rotary extractor. (<i>L2 – Understand, CO3</i>)
Chapter 4: Crystallization	
1.	<ul style="list-style-type: none"> Define crystallization and explain its significance in the purification of chemical products. (<i>L1 – Remember, CO1</i>)
2.	<ul style="list-style-type: none"> Explain the factors affecting crystallization rate and crystal size. (<i>L2 – Understand, CO2</i>)
3.	<ul style="list-style-type: none"> Differentiate between batch and continuous crystallizers with diagrams. (<i>L3 – Apply, CO2</i>)
4.	<ul style="list-style-type: none"> Describe the construction and working of a vacuum crystallizer. (<i>L2 – Understand, CO3</i>)
5.	<ul style="list-style-type: none"> Explain the concept of supersaturation and its role in crystallization. (<i>L2 – Understand, CO2</i>)
6	<ul style="list-style-type: none"> Solve a numerical problem to calculate the yield of crystals from a saturated solution. (<i>L4 – Analyze, CO4</i>)
7	<ul style="list-style-type: none"> Perform a material balance calculation for a crystallization process under steady-state conditions. (<i>L4 – Analyze, CO4</i>)
8	<ul style="list-style-type: none"> Compare the working principles of evaporative crystallization and cooling crystallization. (<i>L3 – Apply, CO3</i>)
9	<ul style="list-style-type: none"> Describe the design considerations for crystallizers used in the pharmaceutical industry. (<i>L2 – Understand, CO3</i>)